

The Structure of the Pyoverdinin from *Pseudomonas fluorescens* 1.3. Structural and Biological Relationships of Pyoverdins from Different Strains*

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The structure of the pyoverdinin siderophore of *Pseudomonas fluorescens* 1.3 was elucidated by spectroscopic methods and chemical degradation. It shows structural similarities with the pyoverdins of several other strains. Whether mutual recognition occurs was investigated by growth tests.

Introduction

The detailed structural investigation of pyoverdins started almost 100 years after Gessard (1892) had first described their formation by fluorescent pseudomonads, namely when Teintze *et al.* (1981) published the structure of pseudobactin. In the meantime over 30 complete or fairly complete structures were elucidated and from preliminary studies it appears that many more are to be expected (Kilz *et al.*, 1999). Pyoverdins consist of three distinct structural parts, viz. a dihydroxyquinoline chromophore responsible for their fluorescence, a peptide chain comprising 6 to 12 amino

acids bound to its carboxyl group, and a small dicarboxylic acid (or its monoamide) connected amidically to the NH₂-group (cf. **1**). Usually several pyoverdins co-occur having the same peptide chain, but differing in the nature of the dicarboxylic acid. The peptide chain has a twofold function. It provides two of the ligand sites for Fe³⁺, and it is responsible for the recognition of the iron complex at the surface of the producing cell (Hohnadel and Meyer, 1988; Budzikiewicz, 1997a). The variability of the peptide chain is closely connected with the second function: It safeguards that a given ferri-pyoverdinin is available only to the producing strain because of a highly specific interaction between the pyoverdinin and a receptor outer membrane protein (Meyer *et al.*, 1979). There are, however, examples known where *Pseudomonas* strains can accept pyoverdins differing in the peptide chain from the one produced by themselves, as certain *P. fluorescens* and *P. putida* strains (Jacques *et al.*, 1995) or *P. aeruginosa* ATCC 15692 that recognizes the pyoverdinin of *P. fluorescens* ATCC 13525 (Hohnadel and Meyer, 1988; Kinzel *et al.*, 1998). It is still an open question whether the foreign ferri-pyoverdinin is accepted by the receptor of the own pyoverdinin due to structural similarities or whether a new receptor is developed (Koster *et al.*, 1993). The first hypothesis is favored by a recent report (Meyer *et al.*, 1999) demonstrating that FpvA, the ferripyoverdinin receptor of *P. aeruginosa* ATCC 15692 efficiently recognizes the pyoverdinin of *P. fluorescens* ATCC 13525. The

Abbreviations: Common amino acids, 3-letter code; OHAsp, *threo*- β -hydroxy Asp; Dab, 2,4-diaminobutyric acid; FoOHOrn, N⁵-formyl-N⁵-hydroxy-Orn; cOHOrn, *cyclo*-N⁵-hydroxy-Orn (3-amino-1-hydroxy-piperidone-2); aThr, *allo*-Thr; TAP-derivatives, N/O-trifluoroacetyl-(amino acid)-isopropyl esters; RP-HPLC, reversed phase high performance liquid chromatography; GC/MS gas chromatograph coupled with a mass spectrometer; ESI, electrospray ionization; FAB fast atom bombardment; HMBC, heteronuclear multiple bond correlation; NOESY, nuclear Overhauser and exchange spectroscopy; ROESY, rotating frame nuclear Overhauser and exchange spectroscopy; TOCSY, total correlation spectroscopy; DSS, [d₆]-2,2-dimethyl-2-silapentane-5-sulfonate; TMS, tetramethylsilane.

* Part LXXXII of the series "Bacterial Constituents". For Part LXXXI see Münzinger *et al.* (1999), for Part LXXX see Voss *et al.* (1999).

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structure elucidation of the *P. fluorescens* pyoverdinin Pf 1.3 (**1**), that shows structural similarities with the pyoverdins of *P. fluorescens* ATCC 17400, *P. fluorescens* 51W, *P. fluorescens* ii and one from a *P. putida* isolate (for the structures see *Discussion*) allows to study the cross uptake between these strains.

Material and Methods

Instruments

Mass spectrometry: Finnigan-MAT HSQ-30 (FAB, matrix thioglycerol/dithiodiethanol), Finnigan-MAT 900 ST (ESI; 50 μ m solutions in CH₃OH/H₂O 1:1 v/v); GC/MS Incos 500 (all Finnigan-MAT, Bremen) with Varian (Sunnyvale CA, USA) GC 3400.

NMR: AM 300 (¹H 300, ¹³C 75.5 MHz) and DRX 500 (¹H 500, ¹³C 125 MHz) (both Bruker, Karlsruhe). Chemical shifts relative to TMS with the internal standard DSS: $\delta(\text{TMS}) = \delta(\text{DSS})$ for ¹H, $\delta(\text{TMS}) = \delta(\text{DSS}) - 1.61$ ppm for ¹³C.

UV/Vis: Lambda 7 (Perkin-Elmer, Überlingen).

Chromatography: RP-HPLC Knauer (Berlin), column Nucleosil 100-C₁₈ (preparative 7, analytical 5 μ m) (Knauer, Berlin); low pressure chromatography column Biogel P-2 (Bio-Rad, Richmond CA, USA), XAD-4 (Serva, Heidelberg); GC: columns SE 54 (CS, Langerwehe), Chirasil-L-Val (Macherey-Nagel, Düren).

Production, isolation and derivatisation of **1**

Pseudomonas fluorescens 1.3 (a fish isolate, Champomier and Richard, 1994) was grown for 72 hrs. in 6 l of a succinate medium (Budzikiewicz *et al.*, 1997). After addition of 10 ml 5% Fe(III)-citrate solution cell material was removed by tangential filtration. The solution was adjusted to pH 5.5–6.0 with 6 M HCl and passed through a XAD-4 resin column. Inorganic material was washed out with 15 l H₂O, then ferri-**1** was eluted with 2.5 l CH₃OH/H₂O 8:2 (v/v), brought to dryness i.v. at 30 °C, redissolved in 0.2 M pyridinium acetate buffer (pH 5.0) and chromatographed on Biogel P-2 with the same buffer. The fraction containing ferri-**1** (detection at 405 nm) was brought to dryness i.v. at 30 °C. For complete removal of the buffer substances were redissolved in H₂O, brought to dryness again and further purified by

RP-HPLC on Nucleosil 100-C₁₈ (solvent 0.1 M CH₃COONH₄-buffer, pH 6.2). Purity was controlled by analytical RP-HPLC. Decomplexation was achieved with 8-hydroxyquinoline (Briskot *et al.*, 1986), free **1** was purified by chromatography on Bio-Gel P-2 with 0.1 M CH₃COOH. For qualitative and quantitative analysis of the amino acids, determination of their configuration and dansyl derivatisation of the free ϵ -amino group of Lys see Briskot *et al.* (1986) and Mohn *et al.* (1990).

Growth stimulation tests

Pyoverdinin-induced growth stimulation tests were performed by plating the bacteria (100 μ l of a 1/10 diluted overnight culture in a rich medium containing of 10 g peptone, 5 g meat extract and 5 g NaCl) in KingB-agar (Difco, Detroit MI, USA) medium supplemented with 500 mg/l of ethylenediamine dihydroxyphenyl acetic acid. UV-sterilized filter paper discs (6 mm diameter) were impregnated with 10 μ l of 1 mM XAD-purified pyoverdinin solutions and applied at the surface of the seeded agar. Bacterial growth was determined after 24 h incubation at 25 °C and scored as (++) for growth diameter ≥ 15 mm, (+) for growth diameter < 15 mm, or (–) for no stimulated growth around discs (Table I).

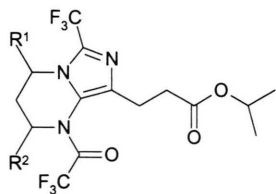
Table I. Growth experiments (for symbols see text).

Strain Producing					Accepts
1	5	8	6	7	
++	+	–	–	–	1
++	++	++	–	–	5
+	+	++	–	–	8
–	–	–	++	+	6
–	–	–	–	++	7

Results

Characterization of **1**

The UV/Vis spectra of **1** (λ 401 nm: pH 6.8; 379 and 368 nm: pH 3.0) and of ferri-**1** (402, 470 and 550 nm: pH 6.8) are typical for pyoverdins (Budzikiewicz, 1997b). The molecular mass as determined by FAB- and ESI-MS is 1285 u. After total hydrolysis the following amino acids could be identified: 1 D-Ala, 1 L-OHAsp, 1 L-Dab, 3 Gly, 1 D-Glu, 1 D-Lys, 1 L-OHOrn, 1 L-Ser plus succinic acid. After dansylation and total hydrolysis ϵ -dan-



2: R¹ = H, R² = COOC₃H₇ or R¹ = COOC₃H₇, R² = H

syl Lys could be isolated showing that the ϵ -amino group of Lys in **1** is free. From the amino acid analysis and the observed molecular mass (as compared with the one calculated for a straight peptide chain, 1822 u) it can be deduced that succinic acid or Glu forms an amide (- 1 u) and that two cyclic substructures are present (- 2 H₂O, i. e. - 36 u).

Sequence determination by NMR

For a detailed discussion of the NMR techniques see Evans (1995). TOCSY allows to detect ⁴J- and ⁵J-coupling within one amino acid residue, NOESY and ROESY allows a correlation between an NH-proton and spatially close α - and β -H of the preceding amino acid (CH-CH-CO-NH-). ²J- and ³J-CH-coupling can be established by HBMC.

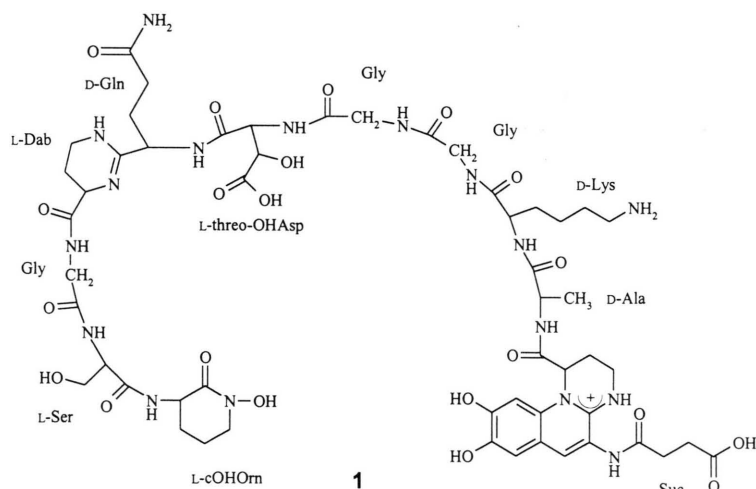
The ¹H- and ¹³C-data are assembled in Tables II and III. Those of the chromophore and of the succinic acid side chain correspond to the ones observed for other pyoverdins (Budzikiewicz, 1997b). The signals of the amino acids including the tetrahydropyrimidine condensation product between Dab and the preceding Gln (esp. the low-field shift of the α -CH at 4.58 ppm; Gwose and Taraz, 1992) and c(OH)Orn (Mohn *et al.*, 1990) could be identified by TOCSY experiments and comparison with literature data. The shift of the CH₂-group of Ser (3.89 ppm) indicates that the OH-group is not esterified (otherwise a downfield shift of about 0.5 ppm would have been expected). The low-field resonance of the Ala-NH (9.60 ppm) is in agreement with the direct connection with the chromophore. The shift values of the ϵ -CH₂-group of Lys agree with the free NH₂-group in accordance with the dansylation experiment (*v. supra*). The presence of Gln rather than Glu is confirmed by the NOESY spectrum (NOE between the amide protons and those of the γ -CH₂-group). Since all amide NH of the peptide chain could be identified (see Table II) the sequence could be determined from the NOESY and ROESY spectra as depicted in Fig. 1. The result is confirmed by the coupling between the carbonyl-C signals of the amide

Table II. ¹H-NMR data (δ [ppm]) of Pf 1.3 (pH 4.3, 5 °C, H₂O/D₂O).

Suc	2'	3'								
	2.70	2.61								
Chr	HN	1	2a	2b	3a	3b	4 HN ⁺	6	7	10
	9.94	5.53	2.56	2.73	3.39	3.76	8.98	7.89	7.01	6.99
	HN	α	β	γ	δ	ϵ	H ₂ N			
Ala	9.60	4.37	1.47							
Lys	8.67	4.24	1.65	1.41	1.45	2.67	7.50			
			1.75	1.21						
Gly 1	8.47	3.88								
Gly 2	8.40	3.95								
		4.12								
(OH)Asp	8.38	4.82	4.51							
Gln	8.94	4.58	2.21	2.48			7.02			
							7.67			
Dab	9.76	4.49	2.15	3.35						
			2.27	3.56						
Gly 3	8.54	4.05								
Ser	8.58	4.48	3.89							
c(OH)Orn	8.67	4.48	1.83	1.95	3.64					
			2.03	2.03	3.70					

Table III. ^{13}C -NMR data ($\delta[\text{ppm}]$) of Pf 1.3 (pH 4.3; 25 °C; D_2O).

Suc	CO (1')	2'	3'	COOH(4')			
Chr	178.2	33.1	33.7	181.7			
	CO	1	2	3	4a	5	
	171.1	57.9	22.9	36.1	150.2	117.6	
	6	6a	7	8	9	10	10a
	140.0	114.9	114.2	145.5	152.7	101.1	133.3
	CO	α	β	γ	δ	ϵ	COOH/CONH ₂
Ala	176.1	51.4	16.8				
Lys	175.8	55.0	31.1	22.8	27.1	40.1	
Gly 1	173.2	43.3					
Gly 2	173.2	43.3					
(OH)Asp	174.3	57.8	72.7				177.5
Gln	164.9	52.8	27.6	31.7			178.0
Dab	173.3	52.4	21.9	37.5			
Gly 3	172.3	43.5					
Ser	172.8	56.6	62.2				
c(OH)Orn	167.8	51.4	27.5	21.0	52.6		



bonds with the following NH- and CH-signals of the following amino acids (Fig. 2).

Mass spectral evidence

In the ESI-MS spectrum of **1** after collision induced fragmentation either in the skimmer region or in the ion trap several sequence-characteristic ions could be observed arising from cleavages of the peptide bonds. They are summarized in Fig. 3 and confirm the conclusions derived from the NMR data. In addition the product of a partial

hydrolysis (6 M HCl, 10 min, 90 °C) was investigated by ESI-MS. The hydrolysis products (Chr... pyoverdinin chromophor) Chr-Ala (m/z 347), Chr-Ala-Lys (m/z 475), [Glu-Dab - H_2O] (m/z 230) and [Glu-Dab - 2 H_2O] (m/z 212) (characteristic for the condensation product giving the tetrahydropyrimidine ring, Gwose and Taraz, 1992), OHAsp-[Glu-Dab - H_2O] (m/z 361) and Gly-OHAsp-[Glu-Dab - H_2O] (m/z 418) could be identified.

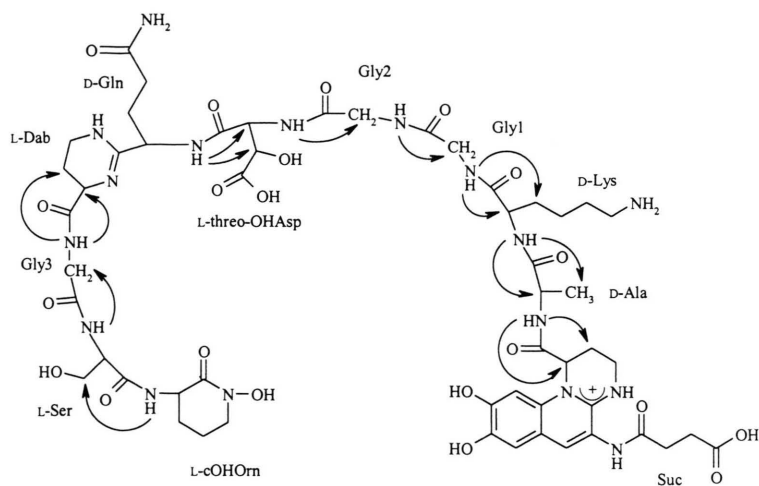


Fig. 1. Sequence specific NOE-cross signals of Pf 1.3 (5 °C, H₂O/D₂O, pH 4.3).

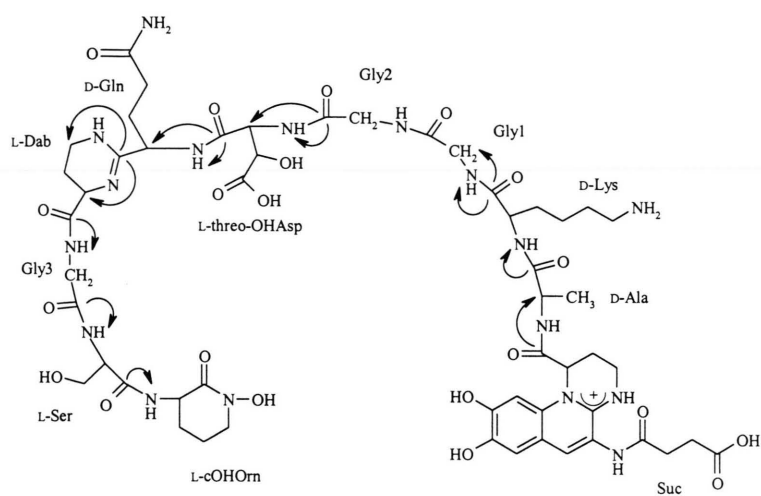


Fig. 2. Sequence specific HMBC cross signals of Pf 1.3 (500 MHz, 5 °C, pH 4.3).

After prolonged hydrolysis (6 M HCl, 40 min, 110 °C) and TAP derivatization (Dallakian and Budzikiewicz, 1997) the derivatives of all amino acids of **1** except those of Dab and Glu could be detected by GC/MS. The dihydropyrimidine ring is generally more stable against acid hydrolysis than amide bonds. Instead, the derivative **2** typical for these condensation products (Filsak *et al.*, 1994) could be observed again confirming the presence of this structural element.

Discussion

The standard opinion is that ferri-pyoverdins are recognized only by the receptor outer membrane protein of the producing strain. As stated in

the *Introduction* exceptions to this rule can be found in the literature. The examples presented so far do not allow any conclusions regarding structural prerequisites for the recognition of a pyoverdinin by a strain that produces a pyoverdinin with another amino acid sequence: From the six siderophores tested by Jacques *et al.* (1995) one (BTP 1) is an isopyoverdinin with a different chromophore, and BTP 7 and BTP 16 are structurally so different (Ongena *et al.*, 1998) that a negative cross-recognition is not surprising. The structures of those pyoverdins where cross-recognition was detected (BTP 2, 7 and 9 as well as BTP 14 accepted by 16) are still unknown. The best-documented (Hohnadel and Meyer, 1988; Kinzel *et al.*,

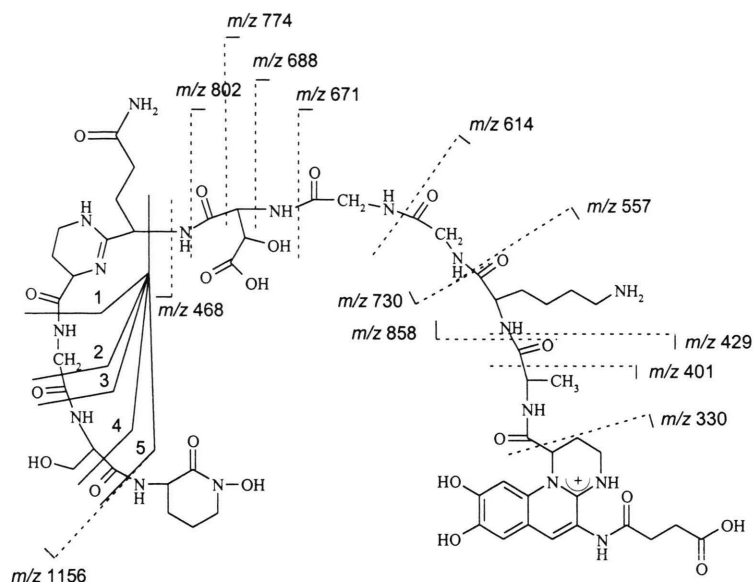


Fig. 3. Characteristic ions in the ESI mass spectrum of Pf 1.3 observed in the ion trap (dotted lines); fragments observed in the MS³ spectrum of m/z 468 (solid lines): 1: m/z 338, 2: m/z 310, 3: m/z 251, 4: m/z 223, 5: m/z 194.

1998; Kinzel *et al.*, 1999) pair are the pyoverdins of *Pseudomonas aeruginosa* ATCC 15692 (**3**) (Briskot *et al.*, 1989) and of *P. fluorescens* ATCC 13525 (**4**) (Hohlneicher *et al.*, 1995). Their common feature is the C-terminal cyclic substructure formed by an amide bond between the carboxyl group of the C-terminal amino acid and the ϵ -amino group of Lys following FoOHOrn (note, that even the first three amino acids show a comparable pattern – D-Ser – basic – small neutral). The similarity of the two structures is obvious. A detailed investigation of the cross-recognition of pyoverdins with a C-terminal macrocycle is on the way. (In the following formulas D-amino acids are underlined, for modified amino acids see *Abbreviations*, Chr indicates the pyoverdine chromophore with the acid side chain. Gln/Dab indicates the tetrahydropyrimidine ring formed by condensation of Gln and Dab as in Fig 1).

Chr-Ser-Arg-Ser-FoOHOrn-(Lys-FoOHOrn-Thr-Thr) (**3**)

Chr-Ser-Lys-Gly-FoOHOrn-(Lys-FoOHOrn-Ser) (**4**)

With the structure elucidation of **1** the mutual recognition of four pyoverdins without a cyclic subunit from *Pseudomonas fluorescens* strains can now be compared which show remarkable structural similarities, viz. **5** from *P. f.* 17400 (Demange *et al.*, 1990), **6** from *P. f.* 51W (Voss *et al.*, 1999) and **7** from *P. f.* ii (Mohn *et al.*, 1990). **8** from *P.*

putida (Gwose *et al.*, 1992) is added to the list because it also contains the tetrahydropyrimidine ring formed by condensation of Gln and Dab following OHAsp.

Chr-Ala-Lys-Gly-Gly-OHAsp-Gln/Dab-Gly-Ser-cOHOrn (**1**)

Chr-Ala-Lys-Gly-Gly-OHAsp-Gln/Dab-Ser-Ala-cOHOrn (**5**)

Chr-Ala-Lys-Gly-Gly-OHAsp-Gln-Ser-Ala-Gly-aThr-cOHOrn (**6**)

Chr-Ala-Lys-Gly-Gly-OHAsp-Gln-Ser-Ala-Ala-Ala-Ala-cOHOrn (**7**)

Chr-Ser-Thr-Ser-Orn-OHAsp-Gln/Dab-Ser-aThr-cOHOrn (**8**)

Recognition of foreign pyoverdins was tested by growth experiments involving the five strains mentioned and their corresponding pyoverdins. As shown in Table I, the best growth stimulation for a given strain was always observed with its own pyoverdine. Those of foreign origin gave variable responses from no effect to growth stimulation which, in some cases, reached an efficiency comparable to the one observed with the homologous pyoverdine.

The N-terminal part of the peptide chains of **1** and **5** is identical; it then follow two small neutral amino acids and the common C-terminal cOHOrn. The strain producing **1** shows the same acceptance for **5** as for **1**, but the strain producing **5** accepts **1** only to a lesser degree than its own pyoverdine.

The second pair are **6** and **7** where again the first six amino acids are identical. The same divergence is observed as for the preceding pair: The strain producing **7** accepts **6** though to a lesser degree than **7**, but the strain producing **6** does not accept **7**.

8 has in common with **1** and **5** OHAsp-Gln/Dab and with **5** the following Ser. The producer of **8** recognizes **5** in the same way as **8**, the producer of **5** accepts **8** and **1**, though to a lesser degree than **5**, and the producer of **1** accepts **8**, but again not as well as **1**.

As had been stated in the *Introduction* that the acceptance of a pyoverdinin produced by a different strain and characterized by a peptide chain different from that of the own pyoverdinin may be due (a) to identical or very similar structures in the part of the peptide chain responsible for the recognition by the own receptor outer membrane protein, (b) by the presence of multiple receptors each being specific to one type of pyoverdinin, or (c) by the induced development of a new receptor. A less specific porin-mediated transport may also be taken into consideration (Meyer, 1992). The growth results reported above seem to show that identical or structurally very similar "key-sequences" can lead to a mutual acceptance, but that the requirements for the recognition are not of the same stringency for every strain, otherwise the

same acceptance would be expected in both directions. The amino acids close to the chromophore apparently are not decisive for the recognition in agreement with the observation that a modification of the basic amino acid (Lys in **4** or Arg in **3**) does not hinder the transport into the cell (Kinzel *et al.*, 1998; Kinzel and Budzikiewicz, 1999). A decision, which of the possible explanations mentioned above for the acceptance of a foreign pyoverdinin actually operates for a given strain, will only be possible when kinetic studies have been performed and the identification of the receptor proteins has been effected. It must also be said that too little is known about the three-dimensional structures of the iron complexes of the ferri-pyoverdins which may play an additional role (information is available only for the solution structures of three ferri-pyoverdins: Mohn *et al.*, 1994; Tappe, 1995; Atkinson *et al.*, 1998).

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